## STUDY OF $\mathrm{Ga}_{2} \mathrm{O}_{3}$ THIN FILMS OBTAINED BY AEROSOL SPRAY PIROLYSIS

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Gallium oxide $\left(\mathrm{Ga}_{2} \mathrm{O}_{3}\right)$, has attracted much attention in recent years as a transparent semiconducting oxide, relatively high electron mobility ( $\sim 150 \mathrm{~cm}^{2} / \mathrm{V} \cdot \mathrm{s}$ ) and due to its ultra-wide band gap ( $\sim 4.9 \mathrm{eV}$ ) [1], become a potential material for high-performance ultraviolet (UV-C) photodetectors in the wavelength range of 200-280 nm [2]. Besides this, $\mathrm{Ga}_{2} \mathrm{O}_{3}$ exists in five crystallographic phases $\alpha, \beta, \sigma, \gamma$ and $\varepsilon$, where only the $\beta$ phase is the most stable at temperatures higher than $900^{\circ} \mathrm{C}$. The crystalline phases can be change from corundum $\alpha-\mathrm{Ga}_{2} \mathrm{O}_{3}$ form at $400^{\circ} \mathrm{C}$ to pure monoclinic $\beta-\mathrm{Ga}_{2} \mathrm{O}_{3}$ increase of the treatment temperature up to $900^{\circ} \mathrm{C}$. Gallium oxide thin films have been grown by various techniques such as chemical vapor deposition (CVD) [3], vacuum thermal evaporation, atomic layer deposition (ALD), but less from chemical solutions. In this research, we have developed the spray pyrolysis technology which is a pretty cheap, low-cost and simple technology, ideal for optoelectronic applications
and allows us to obtain thin films of different thicknesses controlling the temperature and speed of deposition. The obtained film thickness is determined by the rate of precursor solution injection and the duration of deposition process. Thin films of $\mathrm{Ga}_{2} \mathrm{O}_{3}$ films was carried out by aerosol spray pirolysis on Si substrates under an $\mathrm{O}_{2}$ gas flow. Gallium chloride ( $\mathrm{GaCl}_{3}$ ) with molar mass ( 0.5 M ) dissolved in ethanol ( $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}$ ) were used as precursors, they being mixed in an ultrasonic bath for 30 minutes at a temperature of $50-60^{\circ} \mathrm{C}$. Before the deposition process, the solution was left for 24 hours to make the mixture as clean and clear as possible without some sediment. The thermocouple value was the same as the actual temperature of the substrate $450^{\circ} \mathrm{C}$ and the deposition process lasted depending on the desired thickness. The obtained films were characterized by SEM, EDX, optical transparency and electrical conductivity characterization of $\mathrm{Ga}_{2} \mathrm{O}_{3}$ films.

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